

## HEAT CAPACITY OF HIGH-TEMPERATURE SUPERCONDUCTOR (Bi,Pb)<sub>2</sub> Sr<sub>2</sub> Ca<sub>2</sub> Cu<sub>3</sub> O<sub>10+x</sub>

L. Pastor, O. Buchmanová

*Department of Heat Engineering, Faculty of Mechanical Engineering  
Slovak Technical University  
Nám. Slobody 17, 812 31 Bratislava, Slovakia*

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The heat capacity at constant pressure for high temperature superconductor (Bi<sub>0.9</sub>Pb<sub>0.13</sub>)<sub>2</sub>Sr<sub>2</sub>Ca<sub>2</sub>Cu<sub>3</sub>O<sub>10+x</sub> was measured. The data have a peak at T<sub>c</sub> = 107.6 K and the results are in agreement with measurements of other authors.

Heat capacity of high temperature superconductors (HT<sub>c</sub>S) sample with starting composition (Bi<sub>0.9</sub>Pb<sub>0.13</sub>)<sub>2</sub>Sr<sub>2</sub>Ca<sub>2</sub>Cu<sub>3</sub>O<sub>10+x</sub>, prepared by National Research Institute for Materials, Prague [1] was measured in adiabatic calorimeter by steady state method. The mass ratio for HT<sub>c</sub>S sample and parasitic mass (germanium resistance thermometer and electric wires) was approximately 40 : 1. The porosity of this HT<sub>c</sub>S is approximately 15% (the paper [1] gives density 5200 kg/m<sup>3</sup>). For evaluation the following equation was used:

$$c_p = \frac{Ql}{m\Delta T} \quad (1)$$

where  $c_p$  is mean heat capacity for mean temperature  $T = (T_1 + T_2)/2$  and  $\Delta T = T_1 - T_2$ . Each of the steps (for temperature growth from  $T_1$  to  $T_2$ ) was performed with  $\Delta T = 1$  K. Heat capacity vs temperature for this sample, with critical current density [1]  $I_c$  at 77 K and zero magnetic field 1070 A/cm<sup>2</sup> is shown in Fig. 1.

The accuracy of the experiment has been estimated from mean-root-square error, which was evaluated from the relation

$$\delta_c = \frac{1}{m\Delta T} \{ t^2 \delta_Q^2 + Q^2 \delta_t^2 + \left( \frac{Ql}{m} \right)^2 \delta_m^2 + \left( \frac{Cl}{\Delta T} \right)^2 \delta_{\Delta T}^2 \}^{1/2} \quad (2)$$

Measured quantities  $m$ ,  $l$ ,  $\Delta T$ ,  $Q$ , evaluated values for  $c_p$  and the mean-root-square error  $\delta_c$  for six temperatures are shown in the Table 1.

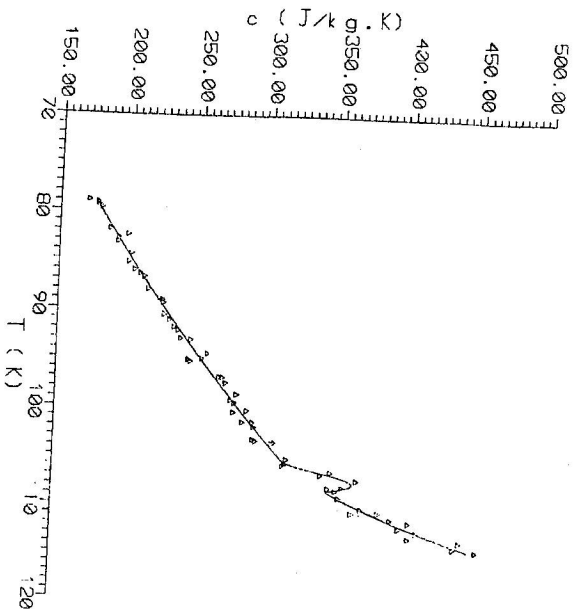


Fig. 1. The heat capacity vs. temperature

Table I. Calculated value  $C$  (from (1)),  $\sigma_c$  (from (2) and measured values  $m$ ,  $l$ ,  $T$ ,  $Q$  for HT<sub>c</sub>S  $\bar{T}$  - average value of measured temperature,  $T_c$  - chosen temperature.

$T$	85 K	95 K	106 K	108 K	110 K	113 K
$\bar{T}$	85.569	94.893	106.189	108.205	110.427	113.246
$m$ [g]	12.050	12.050	12.050	12.050	12.050	12.050
$\sigma_m$	$\pm 0.005$	$\pm 0.005$	$\pm 0.005$	$\pm 0.005$	$\pm 0.005$	$\pm 0.005$
$l$ [s]	32.500	35.250	30.830	22.500	21.700	22.500
$\sigma_l$	$\pm 0.100$	$\pm 0.100$	$\pm 0.100$	$\pm 0.100$	$\pm 0.100$	$\pm 0.100$
$\Delta T$ [K]	2.845	2.496	1.693	1.220	1.18	0.930
$\sigma_{\Delta T}$	$\pm 0.022$	$\pm 0.022$	$\pm 0.022$	$\pm 0.022$	$\pm 0.022$	$\pm 0.022$
$Q$ [mW]	222.133	221.984	225.931	232.583	232.299	212.843
$\sigma_Q$	$\pm 0.025$	$\pm 0.025$	$\pm 0.025$	$\pm 0.025$	$\pm 0.025$	$\pm 0.025$
$\sigma_c$	1.754	2.413	4.5777	6.613	7.694	10.288
$c$	202.679	253.722	325.598	359.628	367.580	421.053
$\bar{c}$ [J/kg/K]	205.401	253.141	331.085	352.968	374.528	425.914

The fitting curves are:

$$c = 44.354 - 1.038T + 0.0034T^2 \quad (3)$$

for the temperature interval (79.2 - 105.8) K, as was determined from 43 measurements, and

$$c = 9.495 \times 10^6 - 2.676 \times 10^5 T + 2.515 \times 10^3 T^2 - 7.8747 T^3 \quad (4)$$

for temperature interval (105.5 - 108) K, determined from 9 measurements, and

$$c = 5.971 \times 10^3 - 1.179 \times 10^2 T + 0.061 T^2 \quad (5)$$

for temperature interval (108 - 114.3) K, calculated from 14 measurements.

The heat capacity vs. temperature reaches its peak at  $T = 107.6$  K (critical temperature  $T_c$  for this HT<sub>c</sub>S is about 108 K [1]). Measurement attested that this sample of HT<sub>c</sub>S has very similar change of the heat capacity vs temperature, like is shown for the polycrystalline sample [2] of BPSCO (2223 phase) with peak at  $T_c = 107$  K. The results entitle to use the same conclusions for fluctuations like is described in Ref. 2. The heat capacity jump  $c$  at  $T_c$  (e.g.  $\Delta c(T_c)/T_c$ ) is approximately 2.5 ml/mol K<sup>2</sup> which is also in agreement with other measurements [2]. The authors prepare the solution for HT<sub>c</sub>S characteristics introduction in which will be respected the theoretical debye function and temperature dependence  $R$  below  $T_c$ .

## REFERENCES

- [1] V. Plecháček, H. Hejlová, Z. Treibalová: Cryogenics 30 (1990) 11
- [2] J. E. Gordon, S. Pridge, S. J. Collocott: Physica C (1991) 185