## THE CHAIN STRUCTURE CONTRIBUTION TO THE MAGNETOVISCOSITY OF A KEROSENE BASED FERROFLUID 1)

## ЦЕПНАЯ СТРУКТУРА ВКЛАДОВ В МАГНИТНУЮ ВЯЗКОСТЬ ФЕРРОЖИДКОСТИ, ОСНОВАННОЙ НА КЕРОСИНЕ

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molecules which reduces the short range Van der Waals interaction. Secondly, the particles are made colloidal stability is achieved in two ways. Firstly, the particles are coated with long-chain surfactant only individual particles but also the clusters of these particles play an important role in physical particles form clusters. From observation of various physical properties, it is concluded that not neighbouring particles in the direction of its magnetic moment. It is expected, therefore, that the as a consequence of the dipole-dipole interaction between particles the particle tends to attract the like water, kerosene, petroleum and so on. These fluids exhibit a superparamagnetic behaviour. The as small as possible in order to reduce the effect of the long-range magnetic dipolar interaction. But Ferrofluids consist of single domain particles (  $\approx 10\,\mathrm{nm}$  in diameter) dispersed in a liquid carrier

cannot explain the anomalous behaviour of viscosity in some ferrofluids [3]. Therefore it seems that  $\Delta \eta / \eta_{\rm olmax} = 3 \varphi / 2$ , where  $\varphi$  is the volume concentration of magnetic particles. This mechanism of rotation particles by the magnetic field. The maximum value of this magnetoviscosity effect is The first suggested mechanism [2] of the viscosity increase was interpreted in terms of the hindrance sider the dependence on the applied magnetic field. The final result in situations when the magnetic dissipation process on the chain structures of the magnetic particles. But Bibik did not conto viscosity was calculated by Bibik [4]. This magnetoviscosity contribution is given by the rather than by the hindrance of rotation of the individual particles. The last mentioned contribution Another possibility is that the magnetoviscosity is caused by the cluster formation of the particles field is perpendicular to the flow of the ferrofluid is given in paper [5] by the formula the magnetoviscosity can hardly be interpreted in terms of the rotation of individual particles The viscosity of magnetic fluids is generally increased by the application of the magnetic field.

$$\Delta \eta(H) = \Delta \eta H / (H^2 + H_B^2)^{1/2}.$$
 (1)

where  $\Delta \eta_B = \mu^2 n/2\pi \mu_0 \gamma R_2^3$  is Bibik's result and  $H_B = \mu/\pi \mu_0 R_2^3$ .  $\mu$  is the magnetic moment of a

Contribution presented at the 8th Conference on Magnetism, KOŠICE, 29. 8.—2.9. 1988

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and large fields the following equations: particles in a chain system and  $\gamma$  is the velocity gradient. Using eq. (1) we obtain in the limits of small particle, n is the concentration of the magnetic particles in unit volume. R, is the distance between

$$\Delta \eta(H \to 0) = \Delta \eta_B H / H_B \tag{2}$$

respectively

and

$$\Delta\eta(H>H_B)=\Delta\eta_B(1-H_B^2/2H^2)$$

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 $H > H_B$  ( $\lg a_x$ ) we can easy obtain using eqs. (2) and (3) the following relation From the slopes of the magnetoviscosity vs magnetic field for  $H \rightarrow 0$  (tg  $\alpha_0$ ) and vs  $1/H^2$  for

$$\lg a_{\infty} / \lg a_0 = H_B^3 / 2.$$
 (4)

which makes it possible to determine  $H_B$  and the  $R_2$  distance between particles in the chain system

of the rotation of individual particles, because the volume concentration is about  $\varphi = 0.03$ experimental points of the viscosity are shown in Fig. 1. The viscosity increase is about twice that of the viscosity in the zero field. It is evident that this increase cannot be explained as a consequence flow times of the ferrofluid with the magnetic field perpendicular to the flow direction. The particles, stabilized with an oleic acid sheath. The values of the viscosity were measured from the For the experimental measurement we have used a kerosene based ferrofluid with Fe<sub>3</sub>O<sub>4</sub>

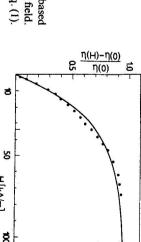
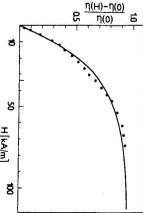


Fig. 1. Magnetoviscosity of a kerosene based The full line is the theoretical fit given by eq. (1) ferrofluid as a function of the magnetic field  $H_B = 34 \text{ kA/m}$ 



value  $H_{\theta}$  for the distance between particles in the chain system the value  $R_2 = 16$  nm. This means  $n = 1.89 \times 10^{22}$  particles/m<sup>3</sup>. If we consider particles with the same size  $D = D_p$ , we can obtain from that the value of the stabilizing sheath of oleic acid on the magnetic particle is about 2 nm when the tion of the ferrofluid is  $I_s = 6.51 \,\mathrm{mT}$  and the concentration of magnetic particles in unit volume is of the parameters of the log-normal distribution we have determined these values: the median particles are in contact. diameter of particles is  $D_s = 12$  nm and the standard deviation  $\sigma = 0.336$ . The saturation magnetizacalculated from room temperature magnetization measurements. The magnetization curve as a function of the magnetic field is given in Fig. 2. Using the Chantrell technique [6] for the calculation the theory was proposed for particles of the same size. The particle size distribution has been between theory and experiment can be explained in terms of the particle size distribution, because experiment is good. The value  $H_B=34\,\mathrm{kA/m}$  has been determined from eq. (4). The discrepancy The theoretical fitting line is given by the full line in Fig. 1. The agreement between theory and

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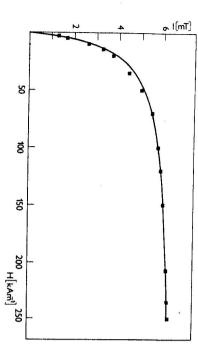


Fig. 2. Room temperature magnetization curve of a kerosene based ferrofluid. The full line is the theoretical fit obtained by the log-normal particle size distribution function and the superparamagnetism theory.

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Accepted for publication February 9th, 1989 Received September 16th, 1988