EPR STUDY OF IMPURITY CENTRES IN NATURAL CaF2

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The EPR method (X-band) was chosen for the identification of impurity ions in a natural single crystal of CaF₂ from Cumberland, England. The presence of Gd³⁺ and Eu²⁺ ions in the EPR spectra of the CaF₂ single crystal was proved as a result of the best fitting procedure. The following values for crystal field parameters were obtained: For Gd³⁺, g = 1.992, $B'_1 = 0.77729 \times 10^{-4}$ cm⁻¹, $B''_1 = 0.00008 \times 10^{-4}$ cm⁻¹, $\Delta E = 0.14916$ cm⁻¹. For Eu²⁺, g = 1.993, $B''_1 = 0.91649 \times 10^{-4}$ cm⁻¹, $B''_1 = -0.00006 \times 10^{-4}$ cm⁻¹, $\Delta E = 0.08707$ cm⁻¹, $^{184}A = 36.0$ cm⁻¹, $^{182}A = 15.1$ cm⁻¹.

ИЗУЧЕНИЕ ЦЕНТРОВ ПРИМЕСЕЙ В ПРИРОДНОМ СаF, С ПОМОЩЬЮ МЕТОДА ЭПР

Для идентификации ионов примесей в природном монокристалле CaF₂, полученном из Кимберленда (Англия). был выбран метод ЭПР (X-диапазон). Присутствие Gd'* и Eu²* в ЭПР-спектрах монокристалла CaF₂ доказано на основе метода наилучшей полгонки. Для параметров внутрикристаллического поля были получены следующие значения: для Gd³*: $g=1,992,\ B_a^0=0,77729\cdot 10^{-4}\ cm^{-1},\ B_a^0=0,01649\cdot 10^{-4}\ cm^{-1},\ AE=0,14916\ cm^{-1},\ для Eu²*: <math>g=1,993,\ B_a^0=0,91649\cdot 10^{-4}\ cm^{-1},\ B_a^0=0,00006\cdot 10^{-4}\ cm^{-1},\ AE=0,08707\ cm^{-1},\ S^1A=36.0\ cm^{-1},\ S^2A=15.1\ cm^{-1}.$

I. INTRODUCTION

Models of paramagnetic centres are suggested on the basis of an analysis of EPR spectra and its angular dependence. All spectra were obtained at room temperatures. The suitability of the models was examined by numeric analysis on a digital computer SIEMENS 4004/150 at the Computing Centre of the Comenius University.

The results indicate that the symmetries of the sites occupied by the Gd³⁺ and Eu²⁺ ions remain cubic. A great many data are available in literature, concerning the iron group and rare earth group ions, both built into synthetic CaF₂ single crystals [1—6] belonging to the O₅ space group, exhibiting cubic [7, 8], tetragonal

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[9—11] or trigonal [12—14] symmetries of the F⁻ ion site. Ca²⁺ ions occupy the sites situated in the centres of the next-neighbour cells.

This paper presents the results of measurements and analyses of the EPR spectra of two impurity ions, namely Gd^{3+} and Eu^{2+} . The electron configuration of the trivalent Gd ion is 4f', the ground orbital singlet state ${}^8S_{7,2}$, the total electron spin of the ion being S=7/2 and the nuclear spin I=3/2. All data are valid for both ${}^{153}Gd$ and ${}^{157}Gd$ isotopes. The electron configuration of the divalent Eu ion is 5f', with the ground state ${}^8S_{7/2}$, which also is an orbital singlet. The total electron spin S=7/2 and the nuclear spin I=5/2 for both natural isotopes ${}^{151}Eu$ and ${}^{153}Eu$.

II. CALCULATION

We succeeded in calculating the energy levels of both Gd^{3+} and Eu^{2+} ions from a Hamiltonian of the following form [4]: $\hat{H}_s = g\beta H_0 \hat{S} + H_0 \hat{G}_s^0 + S\hat{G}_s^0 + H_0 \hat{G}_s^0 + H_0 \hat$

Three different programms were used in order to obtain the best agreement between the measured and calculated positions of the spectral lines and their amplitudes. The first program was used for the best fitting of the crystal field parameters B^m and it was a modified version of the Kalavský and Patráš [15] program. The second, according to [16], was used for the direct calculation of the position of the fine structure spectral lines, as well as of transition probabilities and hyperfine spectral line positions. The last, third program [17] yields the angular dependence of EPR fine structure spectral line positions.

III. EXPERIMENT

The probe crystal was fixed along the direction [0, 0, 1] so that the dc magnetic field vector rotated within the (0, 0, 1) plane when the angular dependences of spectra were measured. Let us note here that the natural cleavage plane of the CaF, single crystal is the (1, 1, 1) plane.

Positions of the centres of spectral lines were measured by means of the proton resonance.

III.1. Gadolinium

The whole EPR spectra of the examined crystal exhibits several tens of spectral lines. However, seven of these lines were distinguishable at first sight because of

used in both cases, which is not the best for ions in the S-ground state [19-22] and it seems to be due to the fact that the Russell—Saunders approximation was values of the intensities is not surprising. A similar discrepancy was reported in [16] respectively. The experiment yields the following intensities of spectral lines: among the electron spin states being -7/2:3/2:5/2:1/2:-5/2:-3/2:7/2, spectra may be expressed as 7:15:12:16:12:15:7, the corresponding transition $imes 10^{-4} \, \mathrm{cm^{-1}}$. The calculated relative intensities of the spectral lines in fine structure $B_n^0 = 0.000008 \times 10^{-4} \text{ cm}^{-1}$ corresponding crystal field parameters are: g = 1.992, $B_4^0 = 0.77729 \times 10^{-4}$ cm⁻¹ environment of the ion into a doublet, a quartet and another doublet. The 1.5:5:4:16:4:11:1.5. The discrepancy between the measured and calculated orbital singlet with an eightfold electron spin degeneracy has been split by the cubic corresponding to the ion in the S-state. The ground state which was originally an crystal field as the LS coupling is $\Delta E = 0.14916 \text{ cm}^{-1}$, which is the value proportional to 16:10:8:1:8:10:17. The zero field energy splitting due to the electron spin states of Gd3+, the separation between the seven lines being line. The main line corresponds to the transition between the -1/2 and +1/2their amplitudes and their symmetric arrangement with respect to the main, biggest $B_4^4 = 3.88645 \times 10^{-4} \text{ cm}^{-1}$ $B_6^4 = -0.00173 \times$

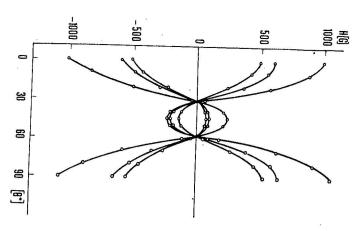


Fig. 1. The angular dependence of EPR spectra of CaF₂: Gd³⁺

The fine structure energy levels of Gd3+ and Eu2+ in CaF2

-7/2 5/2	$\frac{-3/2}{1/2}$ $\frac{-5/2}{3/2}$	1/2 -1/2 -1/2	-5/2 3/2 -3/2	-7/2 5/2	Transition
-5/2 7/2	-1/2 3/2 -3/2	3/2 1/2 1/2	-3/2 5/2 -1/2	-5/2 7/2	ition
4280	3773 3865	2770 3258	2660	2280	The position cal
3972	3858	3250	2543	2105	The position of fine spectral lines calculated Gd ³⁺ Eu ²⁺
4280	3773 3865	2766 3254	2658	2202	Gd ³⁺
3972 4461	3859	²⁶⁸⁶ 3251	2541	2107	exp.

The results of the angular dependence of EPR spectra are plotted in Fig. 1. The excellent agreement between the calculated and measured spectral line positions is seen also in Tab. 1. This agreement was the very argument which led us to the conclusion that one of the impurity ions in the natural CaF₂ single crystal was Gd³⁺.

III.2. Europium

The zero field splitting due to the crystallic field and the LS coupling yields the values $\Delta E = 0.08707 \, \mathrm{cm^{-1}}$, g = 1.993, $B_a^4 = 4.58245 \times 10^{-4} \, \mathrm{cm^{-1}}$, $B_a^6 = 0.00126 \times 10^{-4} \, \mathrm{cm^{-1}}$ for the second impurity ion. In Tab. 1 there are presented the values of line positions calculated by means of the computer, these values being compared with the measured line positions in the EPR spectra, for all the seven lines of the fine structure of Eu²⁺. The agreement is very good. But in the case of the Eu²⁺ ion we have one proof more for its presence in the crystal. It is the well distinguished hyperfine structure presented in Tab. 2. The position of the hyperfine spectral lines were computed and measured from the EPR spectra, exhibiting a good agreement between theory and experiment. The corresponding hyperfine structure constants are: $^{151}A = 36.0$, $^{153}A = 16.1$ in 10^4 cm⁻¹ for the two natural isotopes of 151 Eu and 153 Eu, respectively.

Table 2.

163			2.			2	N)	-3/2→-1/2 2	* ,											5/2→7/2	Transition			
		1	2519			2508	2481	2445	2199	2166	2142		2125		2091		2073		2052	2001	exp.			H
			9	2522.3			2484.5	2447.2	2204.9	2164.0			2123.4	•				2083.4		2004.6	calculated	The	3) line co 4) line co	The hyperfine structure energy levels of the Eu ²⁺ ion on CaF ₂ 1) line corresponds to the transition −5/2→−3/2 2) line corresponds to the transition −3/2→−1/2
	2551.1	2535.0	ı		2518.9	2502.9					2146.4	2129.8		2113.3	2096.8	2084.4	ī	2064.1	2043.8		lated 153Eu	The position of hyperfine spectral lines	 3) line corresponds to the transition 1/2→3/2 4) line corresponds to the transition 3/2→5/2) perfine structure energy levels of the Eu^{2+} ion or 1) line corresponds to the transition $-5/2 \rightarrow -3/2$ 2) line corresponds to the transition $-3/2 \rightarrow -1/2$
¥			3/2→5/2					ls.								-1/2→1/2				÷	Transition	perfine specti	the transition the transition	gy levels of th the transition the transition
10.1	3821	7081	3755	3334	3297	3284	3266	3257	3247	3235	3225		3217	3204	3189			2736 Gd		2717	exp.	ral lines	1/2→3/2 3/2→5/2	e Eu ²⁺ io $-5/2 \rightarrow -3/2 \rightarrow -1$
	3/9/.0	3707 O	3758.3	3346.2	3306.6			3267.4			3228.7				3190.5	3152.7	2773.4	2736.7			calculated			n on CaF ₂ 3/2 1/2
	3816.8	i i				3290.5	3274.1		3257.8	3241.6		v	3225.4	3209.4				0.1777	7771 6	2705.4	lated ¹⁵³ Eu		æ.	9 2
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		3990				3958		٠	3942	3923	2700			Gd			2622		2	2589			2574	2 2556
	3989.2					,	3956.3 ⁴)	3950.0			2698.5			2660.7			2623.4°) 2638.4	363.42	2599.2	2300.2)	2686 211			2560.5
			2712.4	3070 4	3963.1					3930.7 3946.8		2689.2	2673.2		2657.2	2641.3						2583.6	2567.3	
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	4479				4446		į	4422	į	4380 4417		3915.8	9,31,	3917		3903	3888	3010	3979		3861		3840	
						4437.8	4434.2			4364.6 4400 9	N N	ž			3911.3			3875.7	36/3.1)	3973 13)			3836.1	
	27		4466.0	4450.1				7710,5	1118 3				o			3898.7	3882.2				3865.7	3849.3		3833.0

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	4023 4072	4006	
	4028.8 4068.9		
		4012.2	3995.8
4515 4563		4492	
4512.9 4551.1			4475.1
	4498.3	4482.1	

IV. CONCLUSION

 $\Theta = 32^{\circ}$ and 58°, when the dc magnetic field vector was rotated in the (1, 0, 0) spectra in Fig. 1 can serve as the third argument confirming the identification of the impurity ions was correct. All spectral lines diminished to zero at the angles symmetry for the sites occupied by both ions. The angular dependence of the EPR values fulfil the condition $B_a^0 > B_a^0$ proving that the environment has a nearly cubic and B_n^o of Gd^{3+} and Eu^{2+} was performed with standard deviation parameters $K = 0.39192 \times 10^{-4}$ cm⁻¹ and $K = 0.66053 \times 10^{-4}$ cm⁻¹, respectively. The best fit The best fitting procedure in the calculation of the crystal field parameters B'_{ij}

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